

Study the native oxide layer on the surface of semiconductor material GaAs before and after hot-implanted Al ion by **RBS/NR method**

composition of elements in the samples [2].

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MOTIVATION

The electrical characteristics and functionality of semiconductor devices are greatly influenced by this native oxide layer. Studying the formation and growth of the native oxide layer on GaAs surfaces is crucial for improving the performance, reliability, and integration of GaAs-based devices [1]. The GaAs samples were subjected to irradiation with 100 keV-Al ions at a fluence of 4 x 10¹⁶ ions/cm² with various temperatures. Rutherford backscattering spectroscopy with nuclear reaction analysis (RBS/NR) method was used to determine the thickness and atomic

Reference

[1] A.G. Baca, C.I.H. Ashby, Fabrication of GaAs Devices, The Institution of Engineering and Technology, London, 2005 [2] W.K. Chu, J.W. Mayer, M.A. Nicolet, Backscattering Spectrometry, Academic Press, New York, San Francisco, London, 1978. [3] J.R. Cameron, Elastic scattering of alfa-particles by oxygen, Phys. Rev. 90 (1953) 839–844....

Abstract

This work examined the native oxide layer on the GaAs material's surface both before and after hot-implanted aluminum (Al) ions. Rutherford backscattering spectroscopy with nuclear reaction analysis (RBS/NR) method was used to determine the thickness and atomic composition of elements in the samples [2]. The nuclear reaction ${}^{16}O(\alpha,\alpha){}^{16}O$ exhibits elastic resonance at around 3.05 MeV. This resonance provides a useful method for expanding RBS techniques to investigate the concentration of oxygen in oxides [3].

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SAMPLES

For the purpose of research, the GaAs samples were subjected to irradiation with 100 keV-Al ions at a fluence of 4 x 10^{16} ions/cm². Ion implantation performed was at of 25^{0} temperatures C (room temperature), 300⁰ and 500⁰ Celsius.

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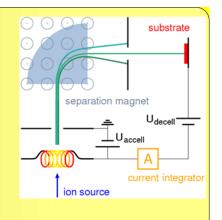
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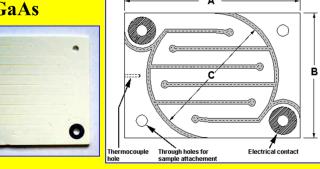
ION IMPLANTATION

Al⁺ implantation was performed with the UNIMAS implanter. The ion beam scanned each sample across a 5 cm square side. The ions current density did not exceed 0.8×10⁻⁶ A cm⁻². Implantation was performed at 23°C, 300°C, and 500°C The dose equal 4×10¹⁶ He⁺/cm² for the each samples. The ion energy was 100 keV.

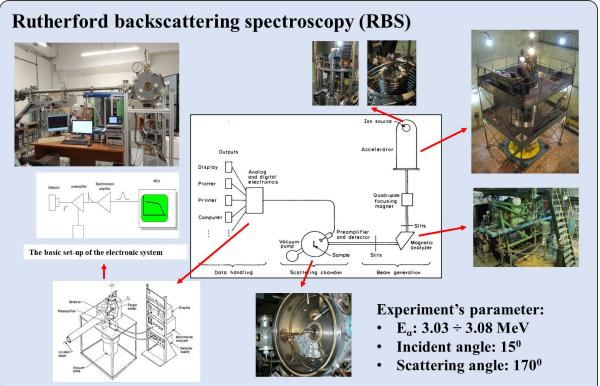


Hot implantation of Al⁺ ions into GaAs

Heater of Samples Boralectric (Tectra) I = 16A, P = 1440 W (Max.)

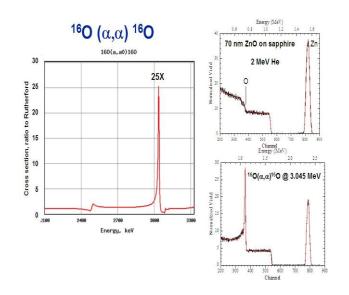


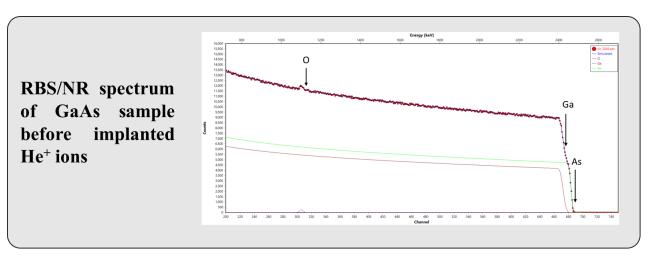
The ceramic heating plate (graphite covered with borium nitride). Max t =800 °C The temperature at the heating holder was stabilized within $+-1^{\circ}C$.

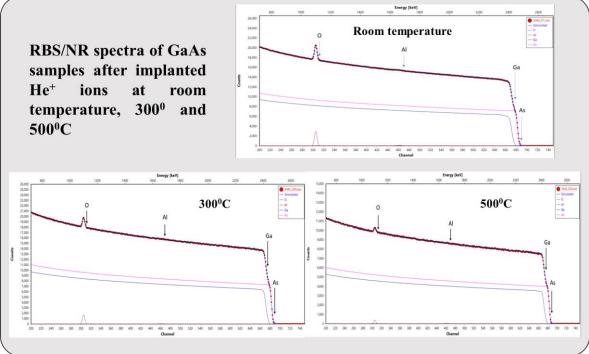


Nuclear reaction analysis (NR)

The nuclear reaction ${}^{16}O(\alpha,\alpha){}^{16}O$ exhibits elastic resonance at around 3.05 MeV. This resonance provides a useful method for expanding RBS techniques to investigate the concentration of oxygen in oxides [**3**]. This represents a distinct resonance that exhibits a backscattering cross-section close the to resonance energy, which is up to times larger than 25 the **Rutherford cross-section.**







Conclusion

- The RBS/NR approach's conclusion indicates that the surface of GaAs samples contains an oxygen-enriched layer.
- The thickness of native oxide layer increase when GaAs samples implanted He⁺ ions at room temperature.
- When the temperature of the Al-implanted process rises, the thickness of this layer decreases.

		Layers	Thickness 10 ¹⁵ atoms/cm ²	Elements concentration			
	Samples			0	Ga	As	Al
	Virgin	1	15	0.33	0.35	0.32	0
		2	20000	0	0.5	0.5	0
	Room temperature (23)	1	20	0.55	0.24	0.2	0.01
		2	40	0.6	0.2	0.16	0.04
		3	200	0	0.46	0.49	0.05
		4	300		0.48	0.5	0.02
		5	20000	0	0.5	0.5	0
	300	1	15	0.5	0.25	0.2	0.05
		2	27	0.36	0.3	0.3	0.03
		3	500	0	0.45	0.5	0.01
		4	20000	0	0.5	0.5	0
	500	1	10	0.5	0.21	0.23	0.06
		2	10	0.35	0.31	0.32	0.02
		3	500	0	0.49	0.5	0.01
		4	20000	0	0.5	0.5	0